UNIVERSITY OF MONTENEGRO INSTITUTE OF FOREIGN LANGUAGES

THE FACULTY OF METALLURGY

ENGLISH COURSE – INTERMEDIATE

INFINITIVE vs. GERUND

THE INFINITIVE

FORM

Depending on the construction, infinitives are used with or without to:
 It's time to go.
 Did you. see the accident happen?

INFINITIVES ARE THEREFORE DIVIDED INTO:

FULL INFINITIVE – TO DO, TO PLAY, TO GO BARE INFINITIVE – DO, PLAY, GO



WE USE BARE INFINITIVE WITHOUT TO:

AFTER VERBS MAKE AND LET:

SHE MADE ME LEAVE.
THEY MAKE ME DO MY HOMEWORK.
SHE HAS MADE ME REPAINT THE HOUSE.

THEY LET US GO.
SHE HAS LET US TAKE IT HOME.

WE ALSO USE BARE INFINITIVE AFTER ALL MODAL VERBS: CAN, SHOULD, MUST, WILL ...



THEY CAN CALL YOU.
WE MUST GO NOW.



WE USE FULL INFINITIVE, WITH TO:

need seem try begin want prefer decide love hate continue like start would like plan forget learn

+ infinitive with to (to do)

advise	agree	choose	decide
ask	beg	tell	invite
expect	hope	offer	promise
plan	refuse	seem	want

I plan <u>to leave</u>. She asked him <u>to stay</u>. He learned <u>to drive</u>.



WE USE FULL INFINITIVE WITH THE FOLLOWING EXPRESSIONS:

I would like

I would love

I would prefer

I would love <u>to help</u> you. They would like <u>to see</u> you.



The gerund

- The gerund is used like a noun: Smoking is bad for you. Do you. like watching TV? She's good at swimming.
- It is formed by adding ing to the infinitive: go -> going stay - staying

like, dislike and other verbs + gerund

Some verbs can be followed by a gerund or a noun, including the following:

like love finish start dislike hate stop enjoy prefer miss give up begin



I prefer/love <u>staying</u> home. Joanna gave up <u>smoking</u>.

COMPARE:

I LOVE SKIING.
I WOULD LOVE TO SKI.

I PREFER CYCLING TO WALKING.
I WOULD PREFER TO STAY HOME.



When a verb follows an preposition, it takes the gerund:
 We thought about leaving early.
 I was worried about getting home.
 I'm interested in hearing more about your offer.
 I'm tired of hearing Ids excuses.
 After closing the door, he looked up and down the street.
 Check your passport before leaving.



PAY ATTENTION:

AFTER ON, IN, AFTER, BY, TO, AT, ABOUT, HOW ABOUT, BEFORE WE USE ING!

THEY ARE TALKING ABOUT (MOVE)

MOVING
SHE IS GOOD AT (SWIM)

SWIMMING

PAY SPECIAL ATTENTION:

AFTER LOOK FORWARD WE USE TO + VERB + ING

SHE LOOKS FORWARD (HEAR) YOU.

TO HEARING

THEY LOOK FORWARD (MEET) US.

TO MEETING

WE ARE LOOKING FORWARD TO (VISIT)
YOU

WE ALSO USE V+ING AFTER EXPRESSIONS:

TO BE BUSY, IT IS NO USE, WOULD YOU MIND

HE IS BUSY PREPARING FOR THE EXAM.

IT IS NO USE TRYING.

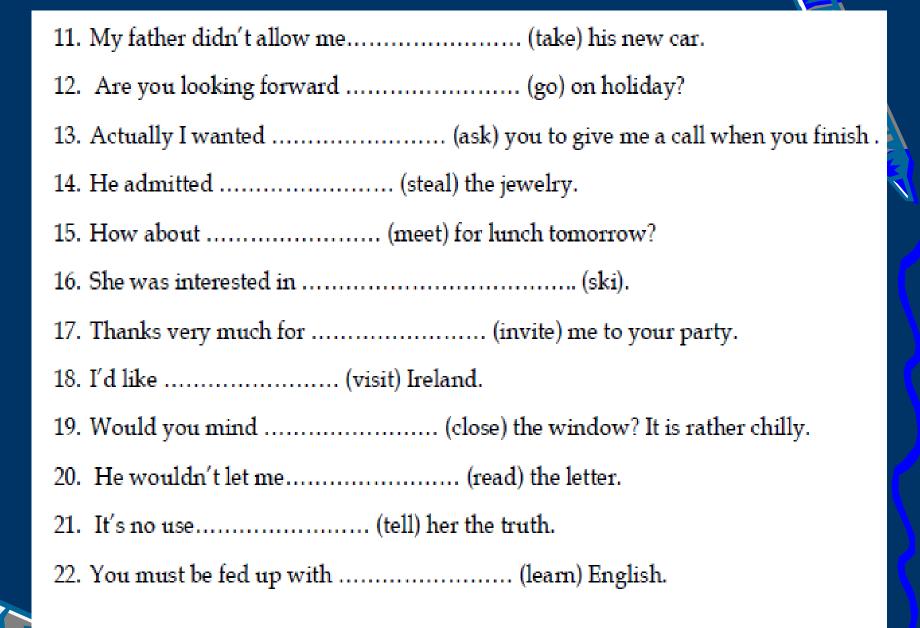
WOULD YOU MIND **OPENING** THE WINDOW?

WOULD YOU MIND DOING THIS EXERCISE?



A. PUT THE VERBS IN BRACKETS INTO THE CORRECT INFINITIVE FORM OR THE -ING FORM:

1. I would like (go out) tonight.	1.		
2. She hates (play) cards.	2.		
3. I can't make a decision. I keep (change) my mind.	3.		
4. He had made his decision and refused (change) his mind.	4.		
5. It was a really good holiday. I really enjoyed(be) by the sea again.	5.		
6. I look forward (hear) from you.	6.		
7. They are bad at (write) essays.	7.		
8. Let us (go) to the theatre tonight.	8.		
9. He is going to ask Ivy (marry) him.	9.		
10. I made him (promise) that he wouldn't tell anybody.			



- I would like TO GO OUT tonight.
- She hates PLAYING cards.
- I can't make a decision. I keep CHANGING my mind.
- He had made his decision and refused TO CHANGE his mind.
- It was a really good holiday. I really enjoyed BEING by the sea again.
- I look forward TO HEARING from you.
- 7. They are bad at WRITING essays. (iza predloga koristimo gerund!)
- Let us GO to the theatre tonight.
- 9. He is going to ask Ivy TO MARRY him.
- I made him PROMISE that he wouldn't tell anybody.

- 11. My father didn't allow me TO TAKE his new car.
- 12. Are you looking forward TO GOING on holiday?
- 13. Actually I wanted TO ASK you to give me a call when you finish.
- He admitted STEALING the jewelry.
- 15. How about MEETING for lunch tomorrow? (iza predloga koristimo gerund!)
- She was interested in SKIING.
- Thanks very much for INVITING me to your party.
- I'd like TO VISIT Ireland.
- 19. Would you mind CLOSING the window? It is rather chilly.
- 20. He wouldn't let me READ the letter.
- 21. It's no use TELLING her the truth.
- 22. You must be fed up with LEARNING English.

Fill in the gaps with the correct form of the verbs in brackets.

- 1 I like *buying* (buy) things on the Internet.
- 2 My boss let me _____ (have) the day off.
- 3 He asked me _____ (help) him move house.
- 4 Would you like ____ (play) tennis?
- 5 My son doesn't mind (wash) the car.
- 6 My parents made me _____ (get up) early.



7	I forgot (learn) the vocabulary.		
8	The number of customers keeps (rise).		
9	Have you told him (come)?		
10	I'd rather (work) at home.		
11	My car wouldn't (start) today.		
12	Liz doesn't need (go) to the meeting.		
13	Should we (try) a different restaurant?		
14	We pay someone (do) the gardening.		
15	I really enjoy (go) to exhibitions.		
16	This seems (be) the right place.		



2 have 3 to help 4 to play 5 washing 6 get up 7 to learn 8 rising 9 to come 10 work 11 start 12 to go 13 try 14 to do 15 going 16 to be



SOLUTIONS

If a lump of sugar is dropped into a beaker of water, it disappears gradually. The sugar is said to dissolve in water. Molecules of sugar become uniformly distributed among the molecules of water. Such a mixture is called a solution. A solution is a homogeneous mixture of two or more substances, the composition of which may vary within characteristic limits. According to their physical state solutions are gaseous like air, solids like the alloy of zinc and copper. The most common solutions, however, are liquids.





The component of a solution that is present in greatest quantity is usually called the *solvent*, the major component, and all other components are called *solutes*.

Certain pairs of substances will dissolve in each other in all proportions. Complete *miscibility* is characteristic of the components of all gaseous solutions and some pairs of components of liquid and solid solutions. For most materials, however, there is a limit on the amount of the substance that will dissolve in a given solvent. The *solubility* of a substance in a

particular solvent at a specified temperature is the maximum amount of the solute that will dissolve in a definite amount of the solvent and produce a stable system.



Substances containing a relatively low concentration of solute are called *dilute solutions;* those of relatively high concentrations are called *concentrated solutions.* A *saturated solution* is when it is in equilibrium with pure solute. An *unsaturated solution* has a lower concentration of solute than a saturated one. A *supersaturated solution* is metastable, that is its solute concentration exceeds the solubility.

Not all substances form true solutions in water. If clay is mixed with water, very little actually dissolves. The result is a heterogeneous mixture called a suspension. Because the components of the mixture have different densities, they readily separate into two distinct phases. Some very small particles, though still much larger than water molecules, are kept permanently suspended. Such mixtures may appear to be homogeneous, but careful examination shows that they are not true solutions. Mixtures of this type are called colloidal suspensions.

Fractional distillation is the separation of the components of a mixture by selective evaporation. For example, to separate a solution of benzene and toluene, we must recondense a part of the distilling vapour and then redistill it. Several of these repetitions of evaporation and condensation will eventually yield pure samples of benzene and toluene.

The terms dilute solution and concentrated solution are qualitative. If it is important to know the ratio of solute molecules to solvent molecules in a solution, it is expressed in molality. *Molality* of a solution is the number of moles of solute per kg of solvent. *Molarity* is the number of moles of solute per liter of solution.

Solutions - Vocabulary

lump – kocka beaker - čaša gradually - postepeno dissolve - rastopiti (se) uniformly - ujednačeno, podjednako homogeneous / gaseous solution - homogeni / gasni rastvor vary - varirati alloy - legura solvent - rastvarač major component - glavni sastojak solute - rastvorak, rastvorena supstanca proportion - odnos, razmera (in all proportions - u svim dovima, u potpunosti)

miscibility - rastvoriljivost, mešljivost solubility - rastvorljivost at a specified temeparature - na određenoj temperaturi stable - stabilan dilute / concentrated solution - razblaženi / koncentrovani rastvor saturated solution - zasićeni rastvor crystallizing - kristalizacija at equal rates - u jednakoj meri unsaturated / supersaturated solution - nezasićeni / presićeni rastvor exceed premašiti, nadmašiti, biti veći od

metastable - metastabilan clay - glina, ilovača suspension - suspenzija density (densities, pl.) - gustina to separate readily - lako se radvajati distinct - razdvojen, odvojen, različit particle - čestica permanently - trajno appear - činiti se, izgledati colloidal suspension - koloidna suspenzija fractional distillation - frakciona destilacija evaporation - isparavanje benzene - benzol, benzen toluene - toluol, toluen

recondence - ponovo kondenzovati vapour - isparenje repetition - ponavljanje yield - dati, dovesti do pure sample - čisti uzorak ratio - razmer, odnos molality - molalnost (broj molova na 1 kg rastarača) molarity - molarnost (broj molova na 1 l rastarača)



A. Answer the following questions:

- What happens if a lump of sugar is dropped into water?
- What is a solution?
- Give an example of a gaseous and solid solutions.
- What is a solvent?
- What is a solute?
- When does a complete miscibility occur?
- What is a saturated solution?
- What is an unsaturated solution?
- What is a supersaturated solution?
- How is a heterogeneous mixture called?



 B. Translate the following expressions into English:

- rastvoriti, rastvor, rastvarač, rastvorljivost
- razblažen rastvor, koncentrovani rastvor, zasićeni rastvor, nezasićeni rastvor, presićeni rastvor
- · mogućnost miješanja, suspenzija



That's all Folks!